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DIRECTED SYNTHESIS OF PHOSPHORUS-CARBON CAGE COMPOUNDS — A CHALLENGE IN ORGANOPHOSPHORUS CHEMISTRY [1]

M. REGITZ*, T. WEITLING, R. FÄSSLER, B. BREIT, B. GEISLER,
M. JULINO, A. HOFFMANN, AND U. BERGSTRÄSSER
Fachbereich Chemie der Universität Kaiserslautern, Erwin-Schrödinger-Strasse,
D-67663 Kaiserslautern, Germany

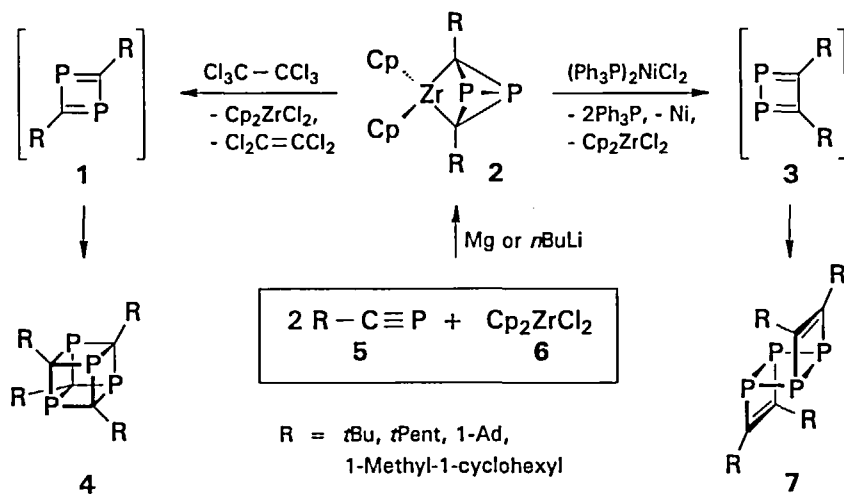
Abstract Reactions of the zirconium complexes **2** with hexachloroethane lead to the tetraphosphacubanes **4** whereas extrusion of the Cp_2Zr units by means of $(\text{Ph}_3\text{P})_2\text{NiCl}_2$ gives rise to the tetraphosphacyclooctadienes **7**. Polycyclic phosphorus-carbon systems such as **11** or **14** and **13** are accessible from multi-step reactions of the phosphaaalkyne **5** ($\text{R} = t\text{-Bu}$) with dienes **9** or tropone (**10**), respectively. The complex **16** obtained from the spirocyclootrimerization of the phosphaaalkyne **5** ($\text{R} = t\text{-Bu}$) with aluminum trichloride provides the starting point for the construction of the bis(homo)prismane **19** and the hexaphosphapentaprismane **20**. Furthermore, the phosphorus-carbon-aluminum cage compounds **12**, **23**, and **24** have been prepared from the phosphaaalkynes **5** and the triorganoaluminum reagents **22**.

INTRODUCTION

In contrast to their all-carbon or all-phosphorus analogs, phosphorus-carbon cage compounds have only become accessible in the past few years and the phosphaaalkynes **5** have proved to be indispensable starting materials. Both purely thermal cyclooligomerization reactions of the latter to furnish tetraphosphacubanes and -cuneanes as well as the corresponding reactions in the presence of organometallic auxiliaries and the coupling reactions of di- and triphosphacyclopentadienides with platinum(II) species are unselective and provide merely modest yields [2]. In this communication, we present reaction strategies for the specific syntheses of phosphorus-carbon polycyclic systems in good to excellent yields.

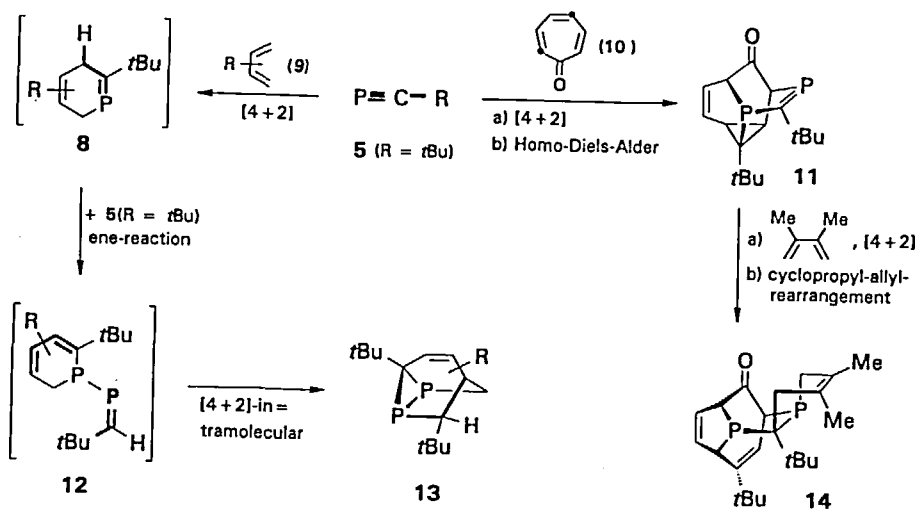
RESULTS

The specific synthesis of tetraphosphacubanes **4** starts from the zirconocene-phosphaaalkyne dimer complexes **2** which, in turn, are accessible from zirconocene dichloride (**6**) and two equivalents of the phosphaaalkyne **5** in the presence of magnesium or *n*-butyllithium [3].



Treatment of the complexes 2 with hexachloroethane effects removal of the zirconocene fragment to afford, after a further dimerization, the pentacyclic product 4 [3]. On the other hand, when the complexes 2 are treated with $(\text{Ph}_3\text{P})_2\text{NiCl}_2$ instead of hexachloroethane, the tetraphosphatricyclooctadienes 7 containing a central P_4 unit are obtained. The 1,3- and 1,2-diphosphetes 1 and 3 are putative intermediates on the way to the polycyclic products [4].

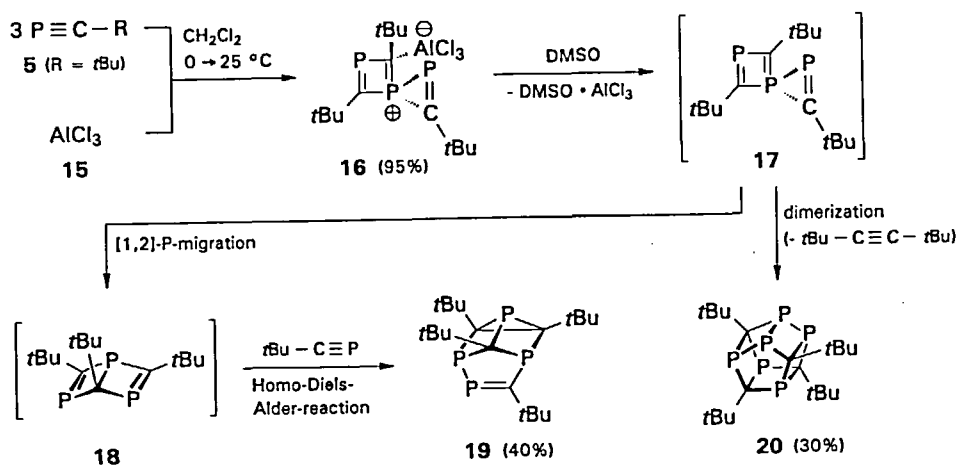
Reaction sequences initiated by Diels-Alder reactions of phosphalkynes have opened up previously unimagined possibilities for the synthesis of phosphorus-carbon cage compounds.



Thus, reactions of the dienophile **5** ($R = t\text{-Bu}$) with variously substituted 1,3-butadienes **9** in a molar ratio of 2:1 furnish the diphosphatricyclooctenes **13** in optimum yields. The reaction mechanism involves an initial $[4 + 2]$ -cycloaddition, an ene reaction with the second equivalent of **5** to give the cyclohexadienylphosphaalkene **12**, and spontaneous isomerization of the latter through an intramolecular Diels-Alder reaction to yield the polycyclic product **13**.

With the same stoichiometry, the initial reaction of **5** ($R = t\text{-Bu}$) with tropone (**10**) affords the Diels-Alder adduct which then reacts with the second equivalent of **5** in a homo-Diels-Alder reaction to provide the diphosphatetracycloundecadienone **11** [6]. The dienophilic properties of such compounds can be exploited for the construction of further polycyclic species (e.g. **11** \rightarrow **14**); in this process the cycloaddition of the 1,3-diene is followed by a sterically initiated cyclopropyl-allyl rearrangement [6].

In the presence of Lewis acids such as aluminum trichloride in a molar ratio of 1:3, the phosphaaalkyne **5** ($R = t\text{-Bu}$) undergoes spirocyclotrimerization to furnish the $1\text{-}\lambda^3\sigma^2, 3\text{-}\lambda^4\sigma^4$ -diphosphete- AlCl_3 adduct **16** [7].

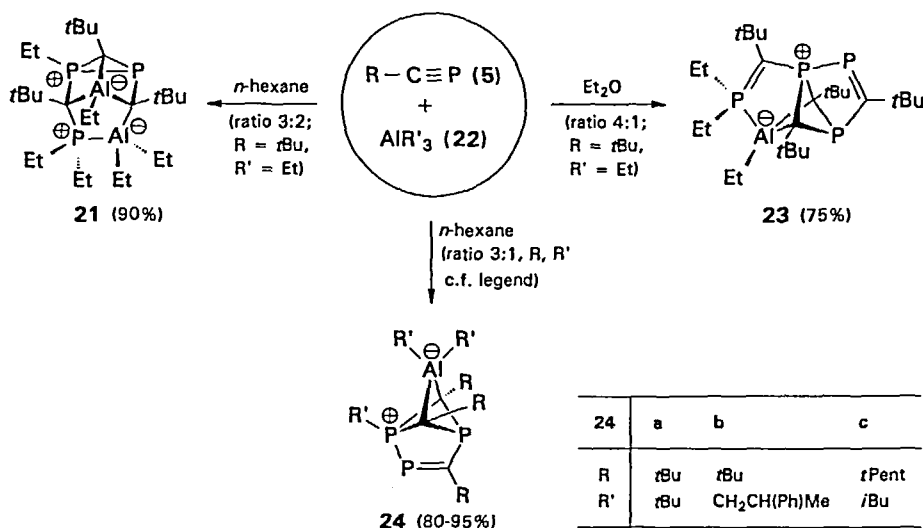


When the more energy-rich spirocyclotrimer **17** is liberated from **16** by treatment with DMSO as a Lewis base, it can only be detected by indirect methods: in dichloromethane at -45°C rearrangement by $[1,2]\text{-P/P}$ migration takes place to furnish the Dewar 1,3,5-triphosphabenzene **18** which, under the prevailing conditions, can only be trapped as **19** after a homo-Diels-Alder reaction with further **5** [7]. In the absence of a trapping reagent, **17** dimerizes with loss of di- t -butylacetylene to the hexaphosphapentaprismane **20** [8].

When phosphaaalkynes are allowed to react with triorganoaluminum reagents, the latter are incorporated into the reaction products; solvent effects and the size of the substituents in the Lewis acid determine the product palette.

In the non-polar solvent n -hexane (molar ratio $5:22 = 3:2$) with moderately large AlR'_3 substituents ($R' = \text{Me}, \text{Et}$), the bis(homo)prismanes **21** with two-fold phosphonium-aluminate character are formed. In the more polar solvent diethyl ether (molar ratio 4:1), the tetracyclic products **23** are produced, again in high selectivity (yields in all cases \geq

75%!) [9]. Voluminous substituents at aluminum (molar ratio 3:1) result in the formation of the triphosphahomobenzvalenes **24** [10].



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